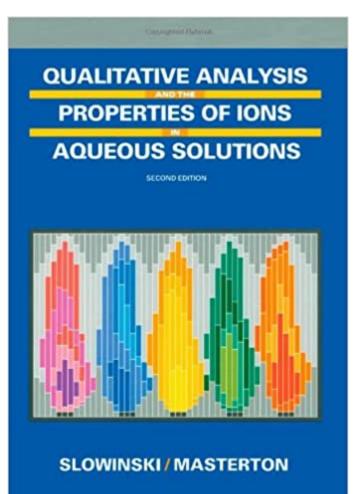


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# Qualitative Analysis And The Properties Of The Ions In Aqueous Solutions (Saunders Golden Series)



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## Synopsis

This inexpensive qualitative analysis supplement offers maximum flexibility and can accompany general chemistry texts. Works well with any general chemistry text, where the instructor wants more qualitative analysis in conjunction with regular class work.

#### **Book Information**

Series: Saunders Golden Series Paperback: 200 pages Publisher: Brooks Cole; 2nd edition (January 2, 1990) Language: English ISBN-10: 0030312345 ISBN-13: 978-0030312342 Product Dimensions: 6.6 x 0.5 x 9.8 inches Shipping Weight: 10.4 ounces (View shipping rates and policies) Average Customer Review: 3.1 out of 5 stars 4 customer reviews Best Sellers Rank: #373,413 in Books (See Top 100 in Books) #111 inà Â Books > Science & Math > Chemistry > Analytic #1286 inà Â Books > Science & Math > Chemistry > General & Reference #1394 inà Â Books > Textbooks > Science & Mathematics > Chemistry

#### **Customer Reviews**

William L. Masterton received his PhD in physical chemistry from the University of Illinois in 1953. Two years later he began to work at the University of Connecticut, where he taught general chemistry and a graduate course in chemical thermodynamics. He has received numerous teaching awards, including an award from the Student Senate at the University of Connecticut, of which he was most proud. Dr. Masterton is co-author of the all-time best-selling general chemistry textbook CHEMICAL PRINCIPLES, which has sold well over 1.5 million copies. Dr. Masterton's field of research, solution thermodynamics, prepared him well for making maple syrup each March at the family farmhouse in New Hampshire.

This book was required for my Chem class in college. I feel that it is difficult to understand until you actually apply the knowledge gained from it. My professor asked that we read the chapter and experiment and write a report on it before we ever performed the actual experiment. The book is very hard to follow, but once the experiments are completed it would start to come together. I appreciate the idea and the detailed explanations in this book, don't know if all of it is necessary.

Exactly what I needed, but there were so many marks in the book, some which I couldn't read over. Only on a few pages.

So many grammar issues that it was distracting. I know it's a chemistry book, but still. I guess it serves its purpose.

Although a little out of date now, this book still contains a wealth of information of value to inorganic chemistry students and TAs alike.

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